## Two Novel Diketopiperazines Isolated from the Fungus Tolypocladium sp.

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Abstract: Diketopiperazines, Sch 54794 and Sch 54796, have been isolated from a fungal fermentation. The structures of these compounds have been established based on spectroscopic data analysis. Sch 54794 exhibited inhibitory activity in the platelet activating factor (PAF) assay.

In the course of searching for bioactive substances from fungal fermentations, 1,2 we have isolated two new compounds, Sch 54794 (1) and Sch 54796 (2), from the fermentation of a fungal culture, *Tolypocladium* sp. The microorganism, *Tolypocladium* sp., was isolated from dead twigs from a *Quercus virginiana* Miller, an old live oak tree in the state of Tamalupas, Mexico.<sup>3</sup> The structures of 1 and 2 were determined as *cis* and *trans* isomers by analysis of the spectroscopic data. This paper details the bioassay-guided isolation and structure elucidation.

The fermentation broth (8 L) was extracted with ethyl acetate at harvest pH. The oily residue from EtOAc extraction (3 g) was dissolved in MeOH-CH<sub>2</sub>Cl<sub>2</sub> (1:1). After removal of insoluble material, the bioactive soluble portion was concentrated and precipitated with hexane. The precipitate was chromatographed by reversed-phase HPLC with aqueous MeOH (YMC-ODS column 30 x 500 mm, 80-90% MeOH in H<sub>2</sub>O with linear gradient in 20 min, 20 mL/min flow rate, UV detection at 210 nm, the active peak  $t_R = 18.5$  min). The active fraction from HPLC was found to be a mixture of two isomers by means of TLC and NMR spectral studies. Attempts to separate the isomers by using silica gel, HP-20P and LH-20 column chromatography with various solvent systems were not successful. It should be noted that these compounds decomposed on the silica gel column, and furthermore, were not separable on either a neutral or basic aluminum oxide



column. The separation of the isomer mixture was successfully accomplished by utilizing a polyvinyl alcohol coated silica gel column with a methanol: 1-chlorobutane solvent system (YMC semi-preparative PVA-SIL column 20 X 250 mm, S-5, 2-10% MeOH in n-BuCl with a linear gradient

in 20 min, 8 mL/min flow rate, UV detection at 275 nm, the first peak, 1,  $t_R = 22.3$  min and the second peak, 2,  $t_R = 25.4$  min). The pure *cis* isomer, 1, (10 mg) and *trans* isomer, 2, (20 mg) were obtained from semi-preparative PVA-SIL column as amorphous white solids after solvent evaporation.

The major component, **2**, mp = 210 °C,  $[\alpha]^{25}D = -25^{\circ}$  (c 0.1, DMSO), was extensively studied by analysis of spectroscopic data for structure elucidation. The molecular weight of **2** was found to be 380 by fast atom bombardment mass spectroscopy (FAB-MS) that showed a protonated molecular ion m/z 381 (M+H)<sup>+</sup>. The elemental analysis of **2** indicated the presence of 2 nitrogen and 2 sulfur atoms. The molecular formula was established as C18H24N2S2O3 by high resolution FAB-MS (Calcd. for C18H25N2S2O3: 381.1307. Found: 381.1302), as well as <sup>13</sup>C NMR spectral data. A very strong peak at m/z 333 in FAB-MS was observed as a (M<sup>+</sup>+H-48) fragment revealing the lose of a methanethiol (CH3SH) unit. The UV spectrum showed absorptions at 277 and 281 nm indicating an O-alkyl benzene moiety. The IR spectrum displayed bands at 3440, 3200 (br. 2NH) and 1670 cm<sup>-1</sup> (amide).

|          | 1                     |                 | 2                    |                 |
|----------|-----------------------|-----------------|----------------------|-----------------|
| Position | <sup>1</sup> H        | 13 <sub>C</sub> | $^{1}\mathrm{H}$     | 13 <sub>C</sub> |
| 2        |                       | 165.5 s         |                      | 165.6 s         |
| 3        | 4.25 (s)              | 58.11 d         | 4.93 (s)             | 57.80 d         |
| 5        |                       | 166.0 s         |                      | 165.9 s         |
| 6        | <i></i>               | 67.62 s         |                      | 68.06 s         |
| 7        | 3.49, 2.97, (d, 13.7) | 44.26 t         | 3.60, 2.95 (d, 13.7) | 42.28 t         |
| 8        |                       | 125.2 s         |                      | 126.1 s         |
| 9        | 7.16 (d, 8.6)         | 131.7 d         | 7.19 (d, 8.7)        | 132.0 d         |
| 9'       | 7.16 (d, 8.6)         | 131.7 d         | 7.19 (d, 8.87)       | 132.0 d         |
| 10       | 6.86 (d, 8.6)         | 114.7 d         | 6.83 (d, 8.7)        | 11 <b>4.6 d</b> |
| 10'      | 6.86 (d, 8.6)         | 114.7d          | 6.83 (d, 8.87)       | 114.6 d         |
| 11       |                       | 158.4 s         |                      | 158.5 s         |
| 12       | 4.50 (d, 6.8)         | 64.70 t         | 4.47 (d, 6.7)        | 64.54 t         |
| 13       | 5.49 (t, 6.8)         | 119.3 d         | 5.46 (t, 6.7)        | 119.4 d         |
| 14       |                       | 138.3 s         |                      | 138.3 s         |
| 15       | 1.76 (s)              | 25.63 q         | 1.74 (s)             | 25.10 q         |
| 16       | 1.82 (s)              | 18.02 q         | 1.79 (s)             | 17.46 q         |
| 17       | 2.24 (s)              | 13.87 q         | 2.22 (s)             | 9.75 q          |
| 18       | 2.30 (s)              | 13.46 q         | 1.48 (s)             | 12.21 q         |

Table 1 <sup>1</sup>H(300 MHz) & <sup>13</sup>C (75 MHz) NMR Data for Sch 54794 (1) and Sch 54796 (2)<sup>a,b</sup>

a. Recorded in CDC13 with small amount of CD3OD, coupling constant (Hz).

b. Assignments are based on APT, COSY and HMBC data.

The <sup>1</sup>H NMR spectral data of **2** which were summarized in Table 1 indicated two vinyl connected methyl singlets at  $\delta$  1.74 and 1.79. The chemical shift of a thiomethyl at  $\delta$  1.48 suggested a strong shielding influence. An oxy-methylene doublet at  $\delta$  4.47 and vinyl proton triplet at  $\delta$  5.46 were observed. Two aromatic proton doublets at  $\delta$  6.83 and 7.19 represented four protons on 1,4-disubstituted benzene ring due to the symmetrical environment. In addition, a methylene group at  $\delta$  2.95 and 3.60 was observed as an AX spin system.

The <sup>13</sup>C NMR spectrum of **2** suggested the presence of a diketopiperazine moiety due to two amide carbonyl signals at  $\delta$  165.6 and 165.9. A quaternary aromatic carbon resonance at  $\delta$  158.5 displayed an oxy-substituted benzene unit. Two vinyl carbon signals at  $\delta$  119.4 and 138.3 along with as oxy-methylene carbon at  $\delta$  64.54 and two methyl carbons at  $\delta$  17.46 and 25.10 were assigned to an isopentenyl group.

The proton and carbon assignments above were supported by COSY experiments. The COSY spectral data showed the oxy-methylene doublet coupled with a vinyl methane triplet and two methyl singlets to form partial structure A. As shown in Figure 1, unit B was established based on the correlation of the AX methylene quartet to the aromatic doublet (H-9, H-9'), which was further coupled to the other aromatic doublet (H-10, H-10'). The remaining part of the molecule was assigned as a trisubstituted diketopiperazine C.



In order to complete the structural assignments for 2, a heteronuclear multiple bond correlation (HMBC) experiment was conducted.<sup>4</sup> The presence of diketopiperazine ring was confirmed by the observation of cross peaks from the H-3 to C-2, C-5 and C-17. Diketopiperazine C and the benzene ring were connected through the methylene group because of the correlation between H-7 and C-2, C-5 and C-6, as well as C-8, C-9 and C-9'. Since the oxy-methylene CH<sub>2</sub>-12 correlated to C-11, the isopentenyl group was obviously linked to the benzene ring at 11-position through a oxygen atom.

The stereochemistry of 2 was established by difference NOE experiments. As depicted in Figure 1 an NOE signal at H-3 was observed when the CH3-18 protons were irradiated. This

evidence revealed the *trans* arrangement of two thiomethyl groups on diketopiperizine ring. Furthermore, the NOE resonance of H-9 and H-10 was also detected with the irradiation of CH<sub>3</sub>-18 protons. This data suggested that the conformation I of 2 was dominating over the conformation II due to a steric hindrance between CH<sub>3</sub>-18 group and aromatic side-chain. Therefore, the structure elucidation of 2 was completed except for the absolute stereochemistry.

The compound 1, mp = 180-182 °C,  $[\alpha]^{25}D = -70^{\circ}$  (c 0.1, DMSO), possessed the same molecular weight as 2 based on FAB-mass spectral data that showed a protonated molecular ion m/z 381 (M+H)<sup>+</sup>. The UV and IR spectra of 1 were also identical to 2. Both <sup>1</sup>H and <sup>13</sup>C NMR data suggested that 1 is an isomer of 2 by direct comparison of their spectra. It should be noted that the up-field chemical shift of H-3 singlet at  $\delta$  4.25 indicated a strong shielding effect in 1. The *cis* isomer of 1 was proposed based on the difference NOE experiment. The only NOE signal of CH3-17 was observed when H-3 singlet was irradiated. This result also revealed that conformation II was more favorable than conformation I because of the lack of a steric hindrance between H-3 and aromatic group.

The *trans* isomer, which is similar to other diketopiperazines reported as PAF inhibitors in literature,<sup>5-7</sup> displayed weak inhibitory activity in PAF assay with an IC<sub>50</sub> value at 50  $\mu$ M. However, the *cis* isomer was found to be inactive (IC<sub>50</sub> >100  $\mu$ M).

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- 3. The fungus was supplied by Dr. B. Katz from MYCOsearch.
- HMBC correlations (<sup>1</sup>H signal---<sup>13</sup>C correlations through 2-bond, 3-bond and 4-bond, respectively): H-3---C-2 (2-bond), C-5, C-17 (3-bond); CH<sub>2</sub>-7---C-6, C-8 (2-bond), C-5, C-9, C-9' (3-bond); H-9---C-10 (2-bond), C-7, C-9', C-11 (3-bond), C-10' (4-bond); H-9'---C-10' (2-bond), C-7, C-9, C-11 (3-bond), C-10 (4-bond); H-10---C-11 (2-bond), C-8, C-10' (3-bond); H-10'---C-11 (2-bond), C-8, C-10 (3-bond); CH<sub>2</sub>-12---C-13 (2-bond), C-11, C-14 (3-bond), C-15 (4-bond); H-13---C-15, C16 (3-bond); CH<sub>3</sub>-15---C-14 (2-bond), C-13, C-16 (3-bond); CH<sub>3</sub>-16---C-14 (2-bond), C-13, C-15 (3-bond); CH<sub>3</sub>-17---C-3 (3-bond); CH<sub>3</sub>-18---C-6 (3-bond).
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